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density, which according to the Born interpretation is given by: Note that in a case such as this, where the wavefunction is purely real, the square modulus of the wavefunction is equivalent to the square of the function. This probability density varies with position within the box. Below the first two wavefunctions for a particle in a box are shown, and below them are corresponding probability densities, as defined by the above equation. This is an example of the correspondence principle, which states that at high quantum numbers, quantum mechanical results reduce to classical mechanical ones. The probability distribution is decidedly nonuniform at low quantum numbers, but as n increases, the probability distribution does become more uniform. The distribution at high quantum numbers is consistent with the classical result that a particle moving between the two walls should, on average, spend equal amounts of time at all positions within the box.

Particle in one-dimensional box problems. Explain the particle in one dimensional box energy quantization. Explain zero point energy of particle in one dimensional box. Application of particle in one dimensional box. Particle in a box. Particle in a box one dimensional. Explain the particle in one dimensional box in detail.